Original Article

In Silico Molecular Docking Againstc-KIT Tyrosine Kinase and ADME Studies of 4-Thiazolidinone Derivatives



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<u>ABSTRACT</u>

Nowadays, the molecular docking approach is used to model the interaction between a small molecule and a protein at the atomic level, which allow us to characterize the behaviour of small molecules in the binding site of target proteins and to elucidate fundamental biochemical processes, C-KIT, a receptor tyrosine kinase, is involved in intracellular signalling, and the mutated form of C-KIT plays a crucial role in the occurrence of some cancers. In this research, we designed novel thiazolidine-4-one derivatives of 3-ethyl-2-(2,4,5-trifluorophenylimino)-thiazolidin-4-one with various benzilidine groups attached to the five-membered imino-thiazolidinone ring and studied molecular docking against C-KIT Tyrosine kinase target protein (1T46). The docking studies of these compounds showed the good interaction of the synthesized molecules with the 1T46 target protein. The ADME studies of these molecules have also been studied to identify which of the synthesized molecules have the potential to cross the Human Intestinal lining (HIA) and the BBB barrier. Out of the 18 molecules studied, 12 derivatives exhibited the good potential to be absorbed by the intestine out of which only one molecule was able to indicate the potential to cross the BBB barrier. There were 5 molecules that could not cross both barriers. These studies could reveal which functionalities present attached to the thiazolidine-4-one could assist in human intestinal absorption and the crossing of the BBB barrier.

Introduction

omputational docking methods are used to screen various possible compounds, searching for new compounds with specific binding properties, or testing a range of modifications of an existing compound. Due to the rapid rise in the amount of molecular biological data available, the computer-aided analysis of molecular interactions becomes more realistic in addition to which as of now the computer prediction of the interaction between proteins and small molecules has advanced to the point that it allows accurate prediction of bound conformations and affinity.

Likewise, the binding of small molecule majorly organic compounds which are ligands to large protein targets is significant to both understanding biological processes and designing drugs [1]. As many proteins regulate biological functions by interacting with small molecules, these receptor proteins are often the prime targets for therapeutic agents. Therefore, a detailed understanding of interactions between small molecules and proteins may form the basis for a rational drug-design strategy which is attractive in drug development concept due to two reasons: it may facilitate the development of more selective therapeutic agents with fewer undesirable side effects and will offer some hopes for reduction of the enormous costs and time required in traditional random screening protocols for drug discovery. Hence, by assuming the receptor structure is available in the PDB database, a major challenge in lead discovery and optimization is to predict both ligand orientation as well as binding affinity which could often be referred to as "molecular docking" [2,3].

Molecular docking has become an increasingly important tool for drug discovery and is the most widely employed technique whose goal is to predict the position and orientation of a ligand (a small molecule) when it is bound to a protein receptor or enzyme. The completion of the human genome project has resulted in broadening the scope of new therapeutic targets in drug design and discovery. Accordingly, the advancement in strategies such as excessive high-throughput protein purification, crystallography, and nuclear magnetic resonance (NMR) spectroscopy has been providing structural information of protein-ligand and protein complexes. This leads to the advancement which resulted in the development of computer-aided drug design, also known as molecular docking [4-5]. Molecular docking is a key tool in structural molecular biology and computer-assisted drug design which tries to

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predict the structure of the intermolecular complex formed between two or more constituent molecules, further trying to predict the position and orientation of a ligand when it is bound to a protein to know the predominant binding modes of a ligand with a protein of known three-dimensional structure. Simply this can be mentioned that docking is a method that predicts the preferred orientation of one molecule to a second when bound to each other to form a stable complex. Usually, binding partners are biological these macromolecules (e.g., protein, DNA/RNA, and peptide) or small molecules (e.g., endogenous ligands and drugs) and their preparations for the docking is just as important as the docking itself [6,7]. The computational approaches are currently being used for screening large databases of compounds to identify potential lead drug molecules. Hence, it can be mentioned that its main application lies in structure-based virtual screening for the identification of new active compounds towards a particular target protein [8]. It can also be stated that for a selected set of structures of a protein and a ligand, the ultimate goal of all docking methods is to predict the structure of the resulting complex and the biological activity of a given ligand.

In this study, molecular docking is performed between receptor i.e. protein molecule and ligand i.e. the novel thiazolidin-4-one derivatives which were already synthesized by the authors. They are the novel thiazolidine-4-one derivatives of 3-ethyl-2-(2,4,5-trifluoro-phenylimino)-thiazolidin-4onewhich belong to an important group of heterocyclic compounds containing sulphur, nitrogen, and carbonyl group in the 4th position in a five-member ring [9,10]. They are an important class of bioactive molecules with diverse biological activities, so it is often called "wonder nucleus". Furthermore, thiazolidinone gives out various derivatives which attracted great attention due to the diversity of their biological effects [11] such as antimicrobial antidiarrheal [12], [13]. [14,15], antidiabetic antiarrhythmic activity[16], anticancer [17-26], anti-HIV [27], Ca²⁺ channel blocker [28], cardioprotective

[29], anti-ischemic [30], cyclooxygenasesinhibitory [31], and anti-platelet activating factor [32].

C-KIT, a receptor tyrosine kinase, is intricate in intracellular signalling, and the mutated form of C-KIT has important role in existence of some cancers. The role of C-KIT has directed to the thought that inhibiting c-Kit kinase activity can be a goal for cancer therapy [33]. The encouraging results of inhibition of c-Kit for treatment of cancers have been detected in some cancers like gastrointestinal stromal tumour, acute myeloid leukemia, melanoma, and other tumours, and these results have stimulated attempts toward improvement of using c-Kit as a capable target for cancer therapy [34]. The main procedure of handling the cancers is chemotherapy, which in anti-tumour compounds are administered to patients. This treatment is thought to be effective. particularly in the early stages of the disease, but it does not permanently cure the patient or totally extinguish cancer. Many factors are associated to the treatment catastrophe, among which we can remark the stage of the disease, the battle of tumour cells to the drugs, and the side effects of the action as the drugs used kill both the cancer cells and the normal cells, often becoming resistant to treatment [35]. It is therefore important to develop effective anti-cancer therapeutic agents with well-defined pharmacokinetic properties.

Therefore, concerning the potentiality of thiazolidinone compounds and CKIT as potential target for cancer theory, we decided to conduct molecular docking against C-KIT Tyrosine kinase target protein (1T46) of the novel thiazolidine-4-one derivatives of 3-ethyl-2-(2,4,5-trifluoro-phenylimino)-thiazolidin-4-onewithvarious benzilidine groups attached to the five-membered imino-thiazolidinone ring.

Experimental

Docking studies

In this study, the affinity and binding modes of the examined molecules against the target protein were determined. First, the water molecules were removed from the crystal structures of target proteins, retaining only main-chain amino acids which are essential for binding. The co-crystallized ligands were used as the reference ligands to predict the binding pockets (Figure 1) and later ligand was removed. Then, the polar hydrogen atoms were added to protein structures to protonate them. The structures of the examined compounds were drawn by using ChemDraw Ultra 7.0 and BIOVIA Discovery Studio Visualizer 2021 which were later saved by using PDB formats. Next, the saved files were opened by using MGL AutoDock Tools software where the protein preparation was done and selected as macromolecule then saved in PDBQT format. The configuration file was created which contained receptor name, ligand name, output file name, X, Y, and Z coordinates of the grid box, and also the size X, Y, and Z of the grid box. Then, the ligand was prepared and any rotatable bonds if available were added. Thereafter, the Command Prompt was opened and AutoDock Vina software was used for running the docking process for each target receptor by ligand by entering necessary codes or commands. In each case, 9 docked structural poses, affinity, and RMSD data were generated by using the algorithm. The output from the Vina split software was further analysed and visualized by using BIOVIA Discovery Studio Visualizer 2021.

Figure 1 depicts the structure of the protein **1T46** on which molecular docking of all 18 earlier synthesized compounds have been performed.

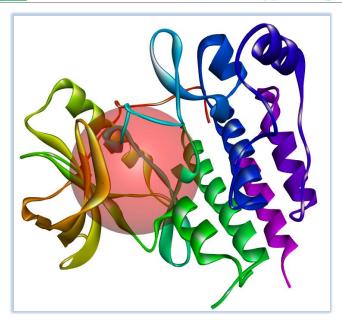


Figure 1. Proteins used for molecular docking - 1T46, C-KIT Tyrosine Kinase target protein

ADME study

ADME study had been performed by using the Swiss ADME site (SwissADME). Primarily the structures were created with the help of ChemDraw Ultra 7.0 software and later

uploaded on the Swiss ADME website to generate Smiles. These Smiles were used to generate ADME analysis data.

Results and Discussion

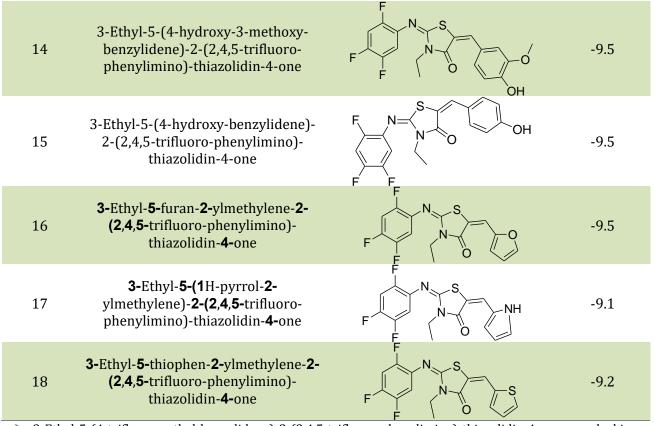
Molecular docking

Molecule No.	Name	Structure	Affinity (kcal/mol)
1	5-Benzylidene-3-ethyl-2-(2,4,5- trifluoro-phenylimino)-thiazolidin-4- one		-10.0
2	3-Ethyl-5-(4-fluoro-benzylidene)-2- (2,4,5-trifluoro-phenylimino)- thiazolidin-4-one		-9.7
3	5-(3-Bromo-4-fluoro-benzylidene)-3- ethyl-2-(2,4,5-trifluoro- phenylimino)-thiazolidin-4-one	$F \xrightarrow{V} V \xrightarrow{V} S \xrightarrow{V} Br$	-7.2
4	5-(2,3-Dichloro-benzylidene)-3- ethyl-2- (2,4,5-trifluoro- phenylimino)-thiazolidin-4-one		-10.2

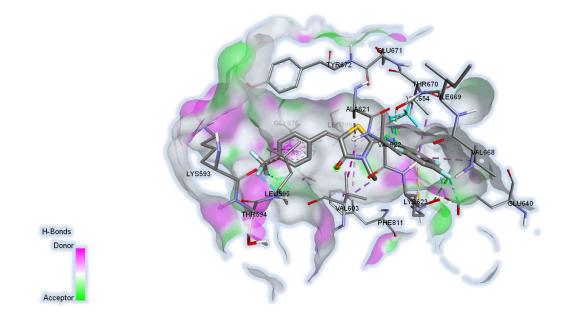
Table 1 Malagulas and their characteristics

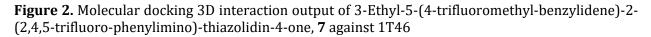
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5	3-Ethyl-5-(2-trifluoromethyl- benzylidene)-2-(2,4,5-trifluoro- phenylimino)-thiazolidin-4-one	$F \xrightarrow{F} V \xrightarrow{S} F \xrightarrow{F} F$	-9.4
6	3-Ethyl-5-(3-trifluoromethyl- benzylidene)-2-(2,4,5-trifluoro- phenylimino)-thiazolidin-4-one	$F \xrightarrow{N} S \xrightarrow{F} F$	-9.0
7	3-Ethyl-5-(4-trifluoromethyl- benzylidene)-2-(2,4,5-trifluoro- phenylimino)-thiazolidin-4-one		-11.1
8	5-(4-Dimethylamino-benzylidene)-3- ethyl-2-(2,4,5-trifluoro- phenylimino)-thiazolidin-4-one		-7.6
9	3-Ethyl-5-(4-fluoro-3-phenoxy- benzylidene)-2-(2,4,5-trifluoro- phenylimino)-thiazolidin-4-one	$F \xrightarrow{F} N \xrightarrow{S} O \xrightarrow{O} O$	-7.8
10	5-(2,3- Dimethoxy-benzylidene)- 3- ethyl- 2-(2,4,5- trifluoro- phenylimino)-thiazolidin- 4- one	F = V + S = O - V + O - O - V + O - O - V + O - O - V + O - O - O - V + O - O - O - O - O - O - O - O - O - O	-9.5
11	5-(3,4- Dimethoxy-benzylidene)- 3- ethyl- 2-(2,4,5- trifluoro- phenylimino)-thiazolidin- 4- one	$F \rightarrow N \rightarrow S \rightarrow O \rightarrow O$	-9.4
12	3-Ethyl-2-(2,4,5-trifluoro- phenylimino)-5-(2,3,4-trimethoxy- benzylidene)-thiazolidin-4-one		-7.1
13	3-Ethyl-2-(2,4,5-trifluoro- phenylimino)-5-(2,4,6-trimethoxy- benzylidene)-thiazolidin-4-one		-6.6



➢ 3-Ethyl-5-(4-trifluoromethyl-benzylidene)-2-(2,4,5-trifluoro-phenylimino)-thiazolidin-4-one docking results against 1T46





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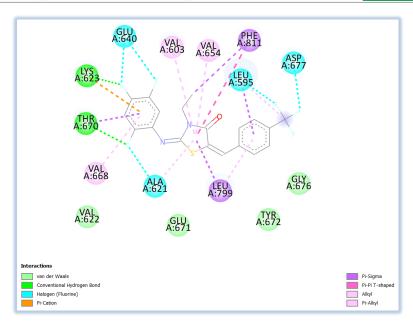


Figure 3. Molecular docking 2D interaction output of 3-Ethyl-5-(4-trifluoromethyl-benzylidene)-2-(2,4,5-trifluoro-phenylimino)-thiazolidin-4-one, 7 against 1T46

To visualize the interactions their 2D diagram and 3D interactions (color of bond interaction

is justified in the table) of H-bonds, i.e. donor and acceptor are shown.

Sr. No.	NAME	COLOUR	DISTANCE	CATEGORY	TYPES OF BONDS	FROM	BONDS	то	BONDS
1	A:LYS6 23:HZ3 - :UNK0:F 24		2.81639	Hydrogen Bond;Halogen	Convention al Hydrogen Bond;Halog en (Fluorine)	A:LYS62 3:HZ3	H- Donor;H alogen Accepto r	:UN K0:F 24	H- Acceptor ;Halogen
2	A:THR6 70:HG1 - :UNK0:F 23		2.56331	Hydrogen Bond;Halogen	Convention al Hydrogen Bond;Halog en (Fluorine)	A:THR6 70:HG1	H- Donor;H alogen Accepto r	:UN K0:F 23	H- Acceptor ;Halogen
3	A:LEU5 95:0 - :UNK0:F 27		3.2834	Halogen	Halogen (Fluorine)	A:LEU5 95:0	Halogen Accepto r	:UN K0:F 27	Halogen
4	A:ALA6 21:0 - :UNK0:F 23		3.61139	Halogen	Halogen (Fluorine)	A:ALA6 21:0	Halogen Accepto r	:UN K0:F 23	Halogen
5	A:GLU6 40:CD - :UNK0:F 24		3.06836	Halogen	Halogen (Fluorine)	A:GLU6 40:CD	Halogen Accepto r	:UN K0:F 24	Halogen
6	A:GLU6 40:OE1 - :UNK0:F 25		2.9049	Halogen	Halogen (Fluorine)	A:GLU6 40:0E1	Halogen Accepto r	:UN K0:F 25	Halogen
7	A:ASP6 77:0D2:		3.11815	Halogen	Halogen (Fluorine)	A:ASP6 77:0D2:	Halogen Accepto	:UN K0:F	Halogen

Table 2. Total Number of Favourable Interactions: 20

2	023, Volume 3	, Issue 1		J	ournal of Ap	oplied Org	anometa	llic Ch	emistry
	В-					В	r	28	
	:UNK0:F								
	28								
0	A:LYS6		4 2 4 2 5 0			A:LYS62	D	:UN	Pi-
8	23:NZ - :UNK0		4.24359	Electrostatic	Pi-Cation	3:NZ	Positive	К0	Orbitals
	A:LEU5								
9	95:CD2		3.7209	Hydrophobic	Pi-Sigma	A:LEU5	С-Н	:UN	Pi-
	- :UNK0					95:CD2		K0	Orbitals
	A:THR6					A:THR6		:UN	Pi-
10	70:CG2		3.43386	Hydrophobic	Pi-Sigma	70:CG2	C-H	K0	Orbitals
	- :UNK0					,01002		110	orbitaib
11	A:LEU7		2 4002	Unduanhahia	Di Ciana	A:LEU7	СIJ	:UN	Pi-
11	99:CD1 - :UNK0		3.4092	Hydrophobic	Pi-Sigma	99:CD1	C-H	K0	Orbitals
	:UNK0:								
10	C14 -		2 74416	TT 1 1 1.	D: C:	:UNK0:C	0.11	A:P	Pi-
12	A:PHE8		3.74416	Hydrophobic	Pi-Sigma	14	C-H	HE8 11	Orbitals
	11								
10	:UNK0 -				Pi-Pi T-		Pi-	A:P	Pi-
13	A:PHE8		5.76464	Hydrophobic	shaped	:UNK0	Orbitals	HE8	Orbitals
	11 :UNK0:				-			11	
	C26 -					:UNK0:C		A:LE	
14	A:LEU5		4.64106	Hydrophobic	Alkyl	26	Alkyl	U59	Alkyl
	95							5	
	:UNK0 -						Pi-	A:LY	
15	A:LYS6		4.58266	Hydrophobic	Pi-Alkyl	:UNK0	Orbitals	S62	Alkyl
	23							3	
16	:UNK0 - A:VAL6		5.40553	Hydrophobic	Pi-Alkyl	:UNK0	Pi-	A:V AL6	Alkyl
10	68		5.40555	liyulophobic	r i-Aikyi	.01110	Orbitals	68	AIKyi
	:UNK0 -						D.	A:V	
17	A:VAL6		4.69753	Hydrophobic	Pi-Alkyl	:UNK0	Pi- Orbitals	AL6	Alkyl
	03				-		Orbitals	03	-
	:UNK0 -						Pi-	A:AL	
18	A:ALA6		3.90841	Hydrophobic	Pi-Alkyl	:UNK0	Orbitals	A62	Alkyl
	21 :UNK0 -							1 A:V	
19	A:VAL6		5.47997	Hydrophobic	Pi-Alkyl	:UNK0	Pi-	AL6	Alkyl
1)	54		5.17 227	nyurophobie	1 1 1111111	.01110	Orbitals	54	11111.91
	:UNK0 -						D:	A:LE	
20	A:LEU7		5.48695	Hydrophobic	Pi-Alkyl	:UNK0	Pi- Orbitals	U79	Alkyl
	99							9	
т	'he st	tructure	of 3-Ft	hvl-5-(4-	distance tvr	nes of hone	te from u	vhoro t	ho hond

The structure of 3-Ethyl-5-(4trifluoromethyl-benzylidene)-2-(2, 4, 5trifluoro-phenylimino)-thiazolidin-4-one as a ligand has been subjected to molecular docking with a protein molecule that would act as a receptor. Docking results observed 9 poses out of which pose having the lowest affinity (kcal/mol) was selected as the best docking pose and was considered for the ligand interaction. Here, 20 favorable interactions were observed where the ligand has bonded at the chosen pocket site in the selected pose. The given table demonstrates the information about the bonds between the ligand and amino acids which contains bond

distance, types of bonds, from where the bond is forming and their types. The ligand formed two hydrogen bonds [conventional hydrogen bond with fluorine], five halogen bonds [with fluorine], one electrostaticbond [pi-cation] and twelve hydrophobic interactions [pisigma,pi-pi T-shaped, alkyl and pi-alkyl].

ADME study

The results of ADME studies of the 18 compounds have been depicted (Figure 4). Molecules with the absorption potential through the intestine appear in the white portion, while the absorbed molecules with

the potential to cross the BBB barrier appear in the yellow portion. The study shows that molecules **1**, **2**, **8**, **10**, **11**, **12**, **13**, **14**, **15**, **16**, **17**, and **18** are capable of being absorbed through the human intestine (HIA). Out of these 12 molecules only molecule **1** is observed to be capable of crossing the BBB barrier. This molecule demonstrates the good permeability potential across the BBB and has no substituents on the phenyl ring attached to the thiazolidin-4-one ring. Likewise, this molecule is demonstrated to follow all five rules of Lipinski's Rule, with no violations of any rule.

Molecule **8** indicates the good potential to be able to cross the BBB barrier, due to the presence of two methyl groups attached to the amine's nitrogen. The presence of heterocycles furan and pyrrole (as in molecules **16** and **17**) are close to the potential molecules that could cross the BBB barrier, while the thiophene presence (as in molecule **18**) decreases its potential to do so, probably due to larger size of sulphur, as compared with oxygen and nitrogen. The molecules **10**, **11**, **12**, **13**, **14**, and **15** having methoxy and hydroxyl groups are the good candidates that could be absorbed by the intestine.

Molecules **5**, **6**, and **7** possessing $-CF_3$ functionalities, and molecule **9** possessing phenoxy group and fluoro groupson the phenyl ring attached to the thiazolidin-4-one ring, show no potential for absorption through the intestinal lining. The presence of halogen atoms -Br and -Cl is indicated to hamper its potential for intestinal absorption, as observed in molecules **3** and **4**. However, if only the -F group is present on the ring, the intestinal absorption improves and demonstrates more potential to cross the BBB barrier, as observed in the case of molecule **2**.

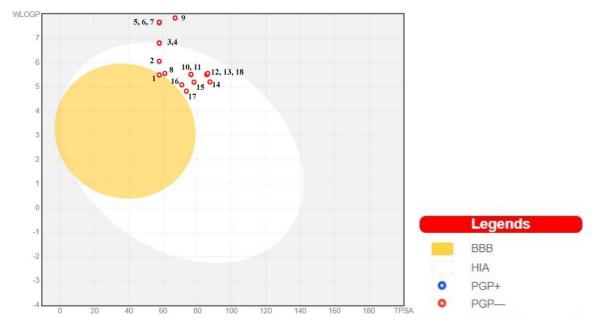


Figure 4. ADME Boiled egg Diagram

Physicochemical Properties									
Molecule No.	Formula	MW	Number of H- bond acceptors	Number of H- bond donors	MR	TPSA			
Molecule 1	$C_{18}H_{13}F_3N_2OS$	362.37	5	0	96.85	57.97			
Molecule 2	$C_{18}H_{12}F_4N_2OS$	380.36	6	0	96.81	57.97			
Molecule 3	$C_{18}H_{11}BrF_4N_2OS$	459.26	6	0	104.51	57.97			
Molecule 4	$C_{18}H_{11}Cl_2F_3N_2OS$	431.26	5	0	106.87	57.97			
Molecule 5	$C_{19}H_{12}F_6N_2OS$	430.37	8	0	101.86	57.97			
Molecule 6	$C_{19}H_{12}F_6N_2OS$	430.37	8	0	101.86	57.97			
Molecule 7	$C_{19}H_{12}F_6N_2OS$	430.37	8	0	101.86	57.97			
Molecule 8	$C_{20}H_{18}F_3N_3OS$	405.44	5	0	111.06	61.21			
Molecule 9	$C_{24}H_{16}F_4N_2O_2S$	472.45	7	0	123.33	67.2			
Molecule 10	$C_{20}H_{17}F_{3}N_{2}O_{3}S$	422.42	7	0	109.84	76.43			
Molecule 11	$C_{20}H_{17}F_3N_2O_3S\\$	422.42	7	0	109.84	76.43			
Molecule 12	$C_{21}H_{19}F_3N_2O_4S$	452.45	8	0	116.33	85.66			
Molecule 13	$C_{21}H_{19}F_3N_2O_4S$	452.45	8	0	116.33	85.66			
Molecule 14	$C_{19}H_{15}F_3N_2O_3S$	408.39	7	1	105.37	87.43			
Molecule 15	$C_{18}H_{13}F_{3}N_{2}O_{2}S$	378.37	6	1	98.88	78.2			
Molecule 16	$C_{16}H_{11}F_3N_2O_2S$	352.33	6	0	89.12	71.11			
Molecule 17	$C_{16}H_{12}F_3N_3OS$	351.35	5	1	91.2	73.76			
Molecule 18	$C_{16}H_{11}F_3N_2OS_2$	368.4	5	0	94.73	86.21			

Table 3. Physicochemical Properties

Table 4. Lipophilicity

			Lipop	hilicity		
Molecule No.	iLOGP	XLOGP3	WLOGP	MLOGP	Silicos-IT Log P	Consensus Log P
Molecule 1	3.45	4.88	5.5	4.12	5.5	4.69
Molecule 2	3.67	4.98	6.06	4.51	5.92	5.03
Molecule 3	3.84	5.68	6.82	5.11	6.6	5.61
Molecule 4	3.87	6.14	6.81	5.11	6.79	5.74
Molecule 5	3.72	5.77	7.67	4.95	6.58	5.74
Molecule 6	3.7	5.77	7.67	4.95	6.58	5.73
Molecule 7	3.73	5.77	7.67	4.95	6.58	5.74
Molecule 8	3.73	5.01	5.56	4.39	5.18	4.78
Molecule 9	4.41	6.51	7.85	5.25	7.04	6.21
Molecule 10	3.97	4.83	5.52	3.83	5.63	4.76
Molecule 11	3.76	4.83	5.52	3.83	5.63	4.71

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Molecule 12	4.08	4.8	5.52	3.5	5.71	4.72
Molecule 13	4.39	4.8	5.52	3.5	5.71	4.79
Molecule 14	3.62	4.5	5.21	3.61	5.09	4.41
Molecule 15	3.07	4.53	5.2	3.54	5.02	4.27
Molecule 16	3.31	3.98	5.09	2.85	4.89	4.02
Molecule 17	3.09	3.71	4.83	2.85	5.02	3.9
Molecule 18	3.43	4.6	5.56	3.7	6.13	4.69

Table 5. Water Solubility

Tuble	5. 114	ter solu	Sincy			Wate	r Solubili	ty				
Molec ule No.	ES OL Log S	ESOL Solubi lity (mg/ ml)	ESOL Solubi lity (mol/l)	ESOL Class	Ali Lo g S	Solubi lity (mg/ ml)	Solubi lity (mol/l)	Class	Silic os- IT LogS w	Silicos -IT Solubi lity (mg/ ml)	Silicos -IT Solubi lity (mol/l)	Silicos- IT class
Molec ule 1	- 5.3 2	1.74E- 03	4.81E- 06	Modera tely soluble	- 5.8 3	5.33E- 04	1.47E- 06	Modera tely soluble	-6.72	6.83E- 05	1.88E- 07	Poorly soluble
Molec ule 2	- 5.4 8	1.26E- 03	3.32E- 06	Modera tely soluble	- 5.9 4	4.40E- 04	1.16E- 06	Modera tely soluble	-6.99	3.88E- 05	1.02E- 07	Poorly soluble
Molec ule 3	-6.4	1.84E- 04	4.01E- 07	Poorly soluble	- 6.6 6	9.99E- 05	2.17E- 07	Poorly soluble	-7.77	7.78E- 06	1.69E- 08	Poorly soluble
Molec ule 4	- 6.5 1	1.32E- 04	3.07E- 07	Poorly soluble	- 7.1 4	3.12E- 05	7.24E- 08	Poorly soluble	-7.9	5.44E- 06	1.26E- 08	Poorly soluble
Molec ule 5	- 6.1 9	2.81E- 04	6.52E- 07	Poorly soluble	- 6.7 6	7.55E- 05	1.75E- 07	Poorly soluble	-7.55	1.20E- 05	2.79E- 08	Poorly soluble
Molec ule 6	- 6.1 9	2.81E- 04	6.52E- 07	Poorly soluble	- 6.7 6	7.55E- 05	1.75E- 07	Poorly soluble	-7.55	1.20E- 05	2.79E- 08	Poorly soluble
Molec ule 7	- 6.1 9	2.81E- 04	6.52E- 07	Poorly soluble	- 6.7 6	7.55E- 05	1.75E- 07	Poorly soluble	-7.55	1.20E- 05	2.79E- 08	Poorly soluble
Molec ule 8	- 5.5 6	1.11E- 03	2.73E- 06	Modera tely soluble	- 6.0 4	3.74E- 04	9.22E- 07	Poorly soluble	-6.8	6.40E- 05	1.58E- 07	Poorly soluble
Molec ule 9	- 6.9 4	5.37E- 05	1.14E- 07	Poorly soluble	- 7.7 2	9.05E- 06	1.92E- 08	Poorly soluble	-9.16	3.25E- 07	6.89E- 10	Poorly soluble
Molec ule 10	- 5.4 8	1.40E- 03	3.33E- 06	Modera tely soluble	- 6.1 7	2.87E- 04	6.79E- 07	Poorly soluble	-6.93	4.95E- 05	1.17E- 07	Poorly soluble
Molec ule 11	- 5.4 8	1.40E- 03	3.33E- 06	Modera tely soluble	- 6.1 7	2.87E- 04	6.79E- 07	Poorly soluble	-6.93	4.95E- 05	1.17E- 07	Poorly soluble
Molec ule 12	- 5.5 6	1.25E- 03	2.76E- 06	Modera tely soluble	- 6.3 3	2.11E- 04	4.67E- 07	Poorly soluble	-7.03	4.22E- 05	9.33E- 08	Poorly soluble
Molec ule 13	- 5.5 6	1.25E- 03	2.76E- 06	Modera tely soluble	- 6.3 3	2.11E- 04	4.67E- 07	Poorly soluble	-7.03	4.22E- 05	9.33E- 08	Poorly soluble
Molec ule 14	- 5.2	2.24E- 03	5.49E- 06	Modera tely	- 6.0	3.58E- 04	8.77E- 07	Poorly soluble	-6.24	2.34E- 04	5.74E- 07	Poorly soluble

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	6			soluble	6							
Molec ule 15	- 5.1 8	2.48E- 03	6.56E- 06	Modera tely soluble	- 5.8 9	4.83E- 04	1.28E- 06	Modera tely soluble	-6.14	2.75E- 04	7.28E- 07	Poorly soluble
Molec ule 16	- 4.6 7	7.48E- 03	2.12E- 05	Modera tely soluble	- 5.1 7	2.36E- 03	6.69E- 06	Modera tely soluble	-5.95	3.99E- 04	1.13E- 06	Modera tely soluble
Molec ule 17	-4.5	1.12E- 02	3.19E- 05	Modera tely soluble	- 4.9 5	3.94E- 03	1.12E- 05	Modera tely soluble	-5.94	4.00E- 04	1.14E- 06	Modera tely soluble
Molec ule 18	- 5.1 6	2.53E- 03	6.87E- 06	Modera tely soluble	- 6.1 4	2.70E- 04	7.33E- 07	Poorly soluble	-5.99	3.77E- 04	1.02E- 06	Modera tely soluble

Table 6. Pharmacokinetics

	Pharmacokinetics								
Molecule	GI abso	BBB	Pgp	CYP1A 2	CYP2C1 9	CYP2C 9	CYP2D 6	CYP3A 4	log Kp
No.	rptio	perme	substr	inhibit	inhibito	inhibit	inhibit	inhibit	(cm/s)
	'n	ant	ate	or	r	or	or	or	
Molecule 1	High	Yes	No	Yes	Yes	Yes	No	No	-5.05
Molecule 2	High	No	No	No	Yes	Yes	No	No	-5.08
Molecule 3	High	No	No	No	Yes	Yes	No	No	-5.07
Molecule 4	High	No	No	No	Yes	Yes	No	No	-4.57
Molecule 5	Low	No	No	No	Yes	Yes	No	No	-4.83
Molecule 6	Low	No	No	No	Yes	Yes	No	No	-4.83
Molecule 7	Low	No	No	No	Yes	Yes	No	No	-4.83
Molecule 8	High	No	No	No	Yes	Yes	No	No	-5.22
Molecule 9	Low	No	No	No	Yes	Yes	No	No	-4.56
Molecule 10	High	No	No	No	Yes	Yes	No	No	-5.45
Molecule 11	High	No	No	No	Yes	Yes	No	No	-5.45
Molecule 12	High	No	No	No	Yes	Yes	No	No	-5.65
Molecule 13	High	No	No	No	Yes	Yes	No	No	-5.65
Molecule 14	High	No	No	No	Yes	Yes	No	No	-5.6
Molecule 15	High	No	No	No	Yes	Yes	No	Yes	-5.39
Molecule 16	High	No	No	Yes	Yes	Yes	No	Yes	-5.62
Molecule 17	High	No	No	Yes	Yes	Yes	No	Yes	-5.81
Molecule 18	High	No	No	Yes	Yes	Yes	No	No	-5.28

Table 7. Druglikeness

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			Dru	ıglikeness		
Molecule No.	Lipinski Violatio ns	Ghose Violations	Veber Violations	Egan Violations	Muegge Violations	Bioavailability Score
Molecule 1	0	0	0	0	0	0.55
Molecule 2	1	1	0	1	0	0.55
Molecule 3	1	1	0	1	1	0.55
Molecule 4	1	1	0	1	1	0.55
Molecule 5	1	1	0	1	1	0.55
Molecule 6	1	1	0	1	1	0.55
Molecule 7	1	1	0	1	1	0.55
Molecule 8	1	0	0	0	1	0.55
Molecule 9	1	1	0	1	1	0.55
Molecule 10	0	0	0	0	0	0.55
Molecule 11	0	0	0	0	0	0.55
Molecule 12	0	0	0	0	0	0.55
Molecule 13	0	0	0	0	0	0.55
Molecule 14	0	0	0	0	0	0.55
Molecule 15	0	0	0	0	0	0.55
Molecule 16	0	0	0	0	0	0.55
Molecule 17	0	0	0	0	0	0.55
Molecule 18	0	0	0	0	0	0.55

Table 8. Medicinal Chemistry

		I	Medicinal Chemistry	
Molecule No.	PAINS Alerts	Brenk Alerts	Leadlikeness Violations	Synthetic Accessibility
Molecule 1	0	3	2	3.7
Molecule 2	0	3	2	3.7
Molecule 3	0	3	2	3.72
Molecule 4	0	3	2	3.73
Molecule 5	0	3	2	3.83
Molecule 6	0	3	2	3.8
Molecule 7	0	3	2	3.8
Molecule 8	1	3	2	3.93
Molecule 9	0	3	2	4.03
Molecule 10	0	3	2	3.93
Molecule 11	0	3	2	3.87
Molecule 12	0	3	2	4.08
Molecule 13	0	3	2	4.07
Molecule 14	0	3	2	3.77
Molecule 15	0	3	2	3.67
Molecule 16	0	3	2	3.68
Molecule 17	0	3	2	3.71
Molecule 18	0	3	2	3.67

Conclusion

The synthesized thiazolidin-4-one derivatives have revealed the good interactions with the C-KIT Tyrosine Kinase (1T46) target protein, which indicates the good anti-viral potential of the molecules. The ADME studies show that groups like fluoro, hydroxyl, methyl, and methoxy demonstrate the good human intestinal absorption, which is not the case if chloro, bromo, and trifluoromethyl groups are attached to the heterocyclic ring. The in-silico methods can illustrate which functional groups in the molecules could aid in absorption in the body. This could give a proper direction to the synthetic organic chemist for synthesizing bio-active derivatives of thiazolidin-4-one.

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